

Orígín of Elastícíty

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EG1101 – Mechanical Engineering – Mechanics of Materials





Linear elastic behaviour

- Ball and spring model of elasticity:
- Material in tension bonds stretched and atoms move apart
- Material in compression bonds compressed and atoms move closer together

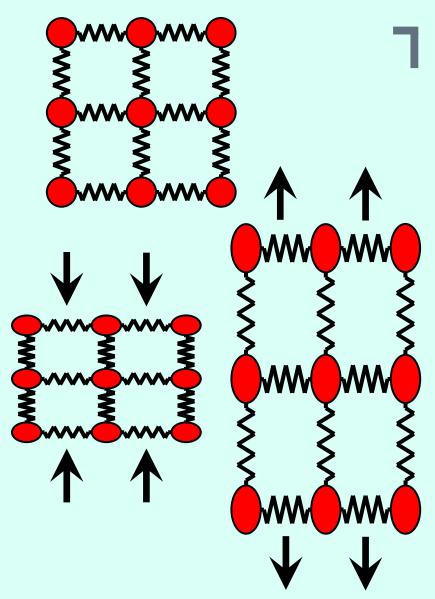


Illustration adapted from: Gordon, J.E., (1978), Structures or Why Thing's Don't Fall Down



Bonding energies

Ball and spring model:

Atoms subjected to competing repulsive and attractive forces

At rest, atoms will be in their lowest energy state

r₀ = equilibrium interatomic spacingU = potential energy of bondr = interatomic spacing

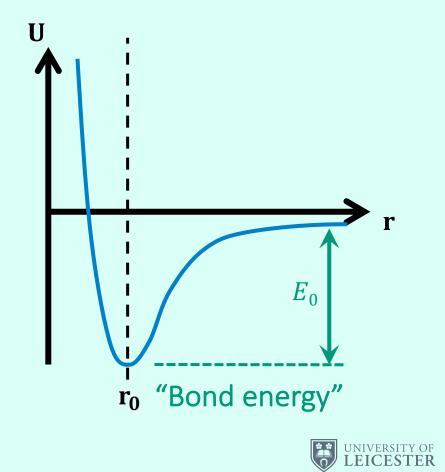
Interatomic spacing, r **Total bond energy** IJ **Energy of** repulsion Energy of attraction ro Described by: Lennard-Jones, **Potential Energy Curve**

Bonding energies



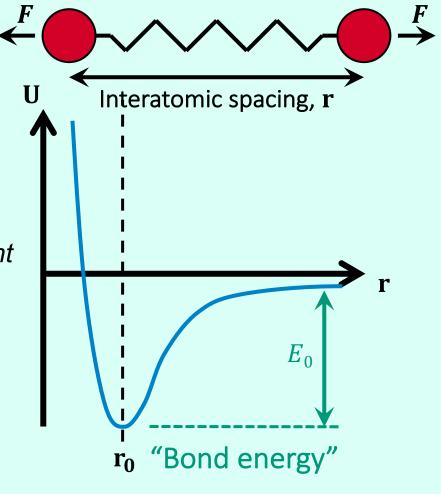
Interatomic spacing, r

- Bond energy (E₀) is the minimum of potential vs spacing curve
- Bond energy describes energy required to break the bond between (dissociate) atoms
- Depth of the 'well' in the total energy curve gives bond strength:
 - Deep well = strong
 - Shallow well = weak



What do bonding energies mean in terms of materials properties?

- Greater bond energy:
 - Greater stiffness (*E*)
 - Greater strength (σ)
 - Higher melting point (*T_m*)
 Lower thermal expansion coefficient (α)



*Note that there are A LOT of exceptions to these generalisations!







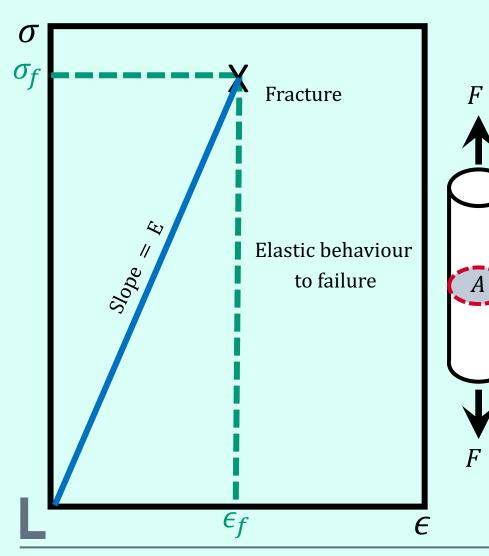
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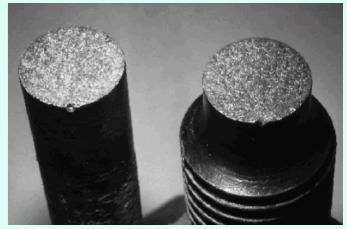
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Deformation – Brittle Fracture





Brittle fracture in mild steel



Indirect tensile test of a concrete beam https://expeditionworkshed.org/workshed/tensilefailure-of-ordinary-concrete-brazilian-test/



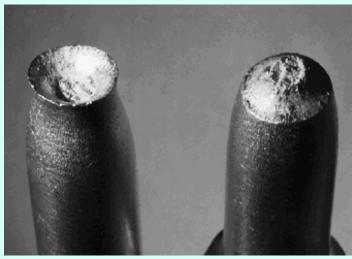
Deformation

Brittle materials

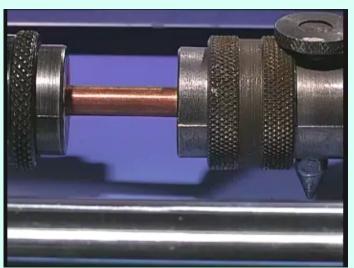
- In brittle materials, defects or flaws control the strength. Hence the material is weak in tension but strong in compression.
- Little/no dislocation movement
- Glass, ceramics, some metals
- Brittle materials lack toughness



Deformation – Ductile Fracture

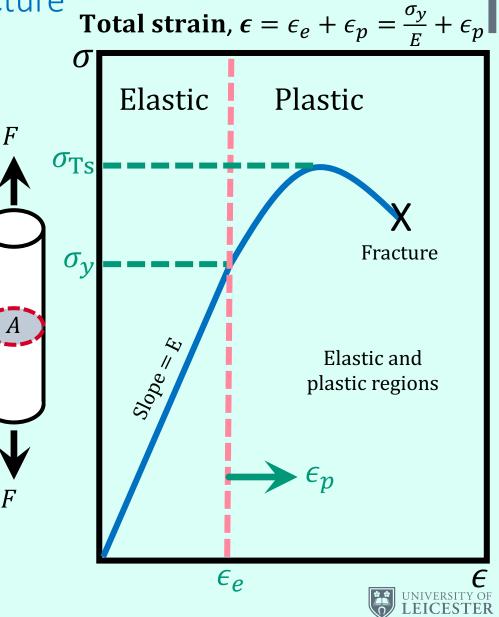


Cup-and-cone fracture in Al



F

Ductile fracture of annealed copper, Courtesy of DoITPoMS, The University of Cambridge



Plastic Deformation

- Shear stresses or direct stresses, depending on the way you look
- Dislocations are a misalignment of atoms in a material.
- In some materials, dislocations can move easily.
 - This is called slip and is the mechanism responsible for plastic deformation.
- In a ductile material dislocations can move easily allowing plastic strain to develop above the yield stress of the material.

